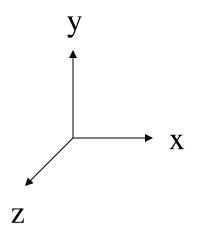
## More on:

Molecular Orbitals

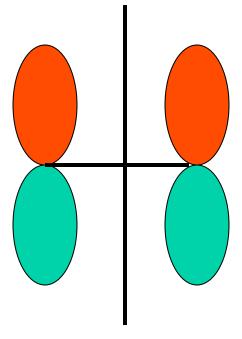
Pericyclic Reactions

Electrocyclic Reactions

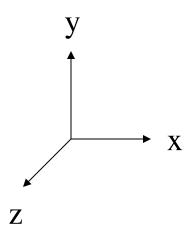


Pi Bonding MO (bond along x-axis; orbitals in xy plane)

mirror in yz plane

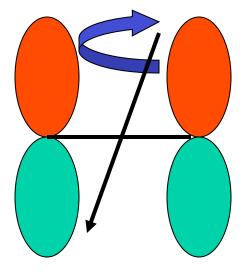


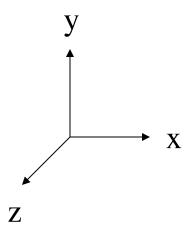
Symmetrical mirror plane (yz)



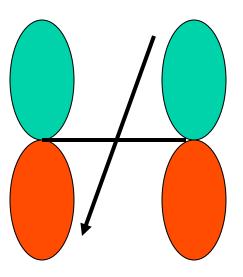
Pi Bonding MO (bond along x-axis; orbitals in xy plane)

rotate 180° about z-axis in xy-plane

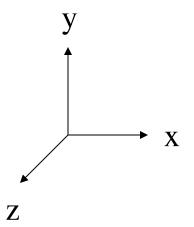




Pi Bonding MO (bond along x-axis; orbitals in xy plane)

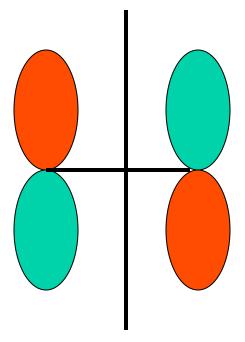


Antisymmetric about the  $C_2$  axis

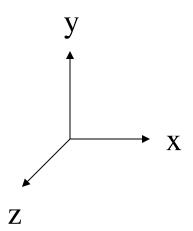


Pi Antibonding MO (bond along x-axis; orbitals in xy plane)

mirror in yz plane

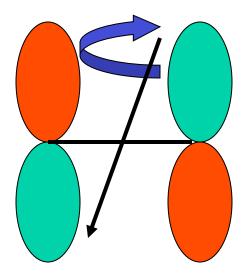


Antisymmetric mirror plane (yz)



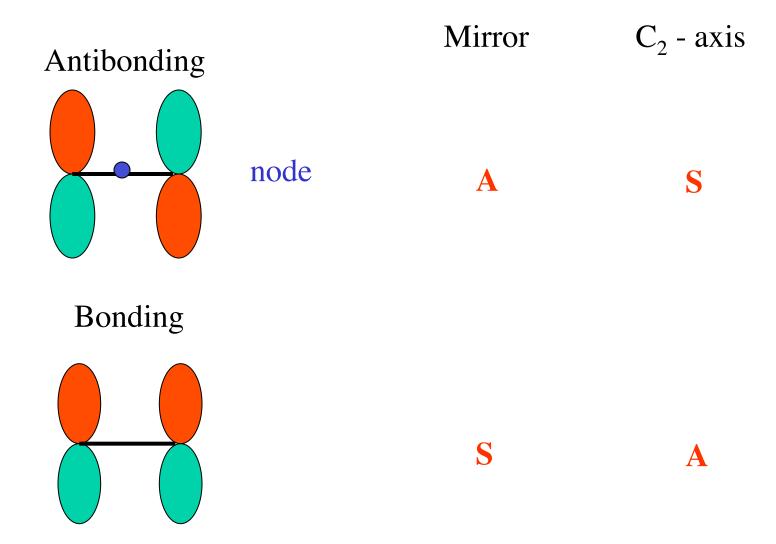
Pi Bonding MO (bond along x-axis; orbitals in xy plane)

rotate 180° about z-axis in xy-plane



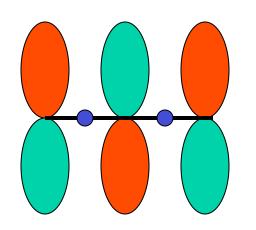
Symmetric about  $C_2$  axis

#### Pi Bond MO

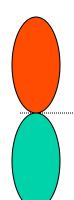


### Allylic Resonance

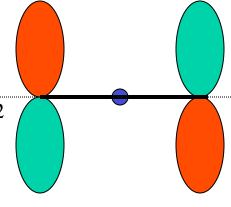
anti-bonding -  $\pi_3^*$ Energy of ean isolated p-orbital



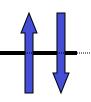
SA —



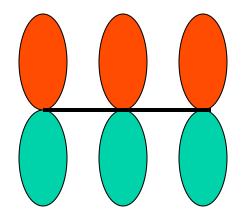
non-bonding -  $\pi_2$  1 node



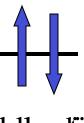
AS



bonding -  $\pi_1$  0 nodes



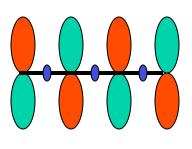
SA



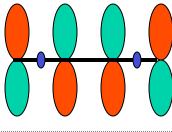
adlyllrantion

### M.O.s of 1,3-Butadiene

anti-bonding M.O. -  $\pi_4^*$ Energy of isolated p-atomic orbital (A.O.) anti-bonding M.O. -  $\pi_3^*$ LUMO



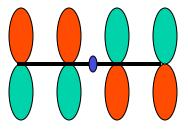
AS



SA

Lowest
Unoccupied
Molecular
Orbital
Highest

bonding M.O. -  $\pi_2$  HOMO

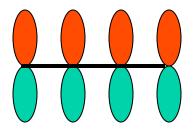


AS

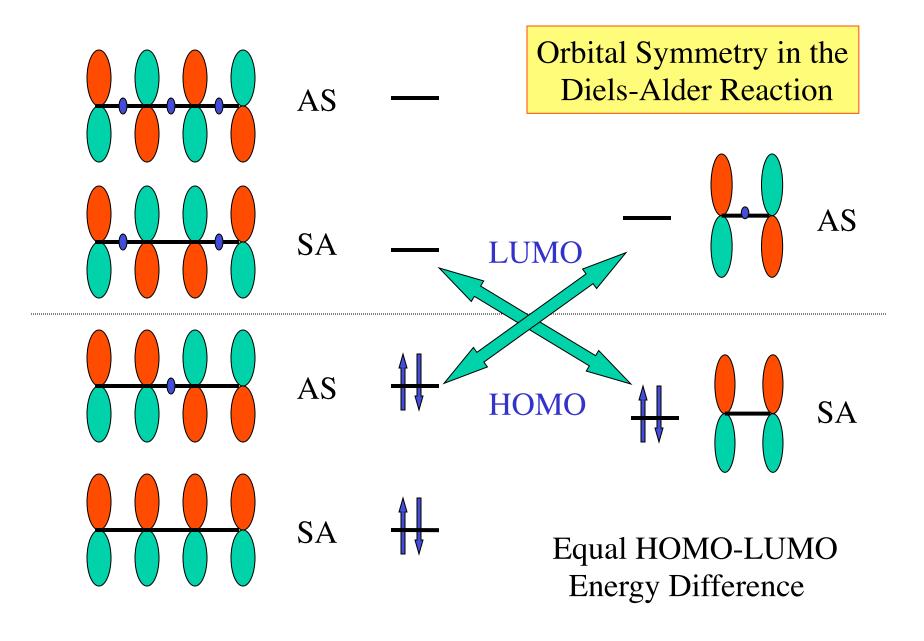
Occupied Molecular

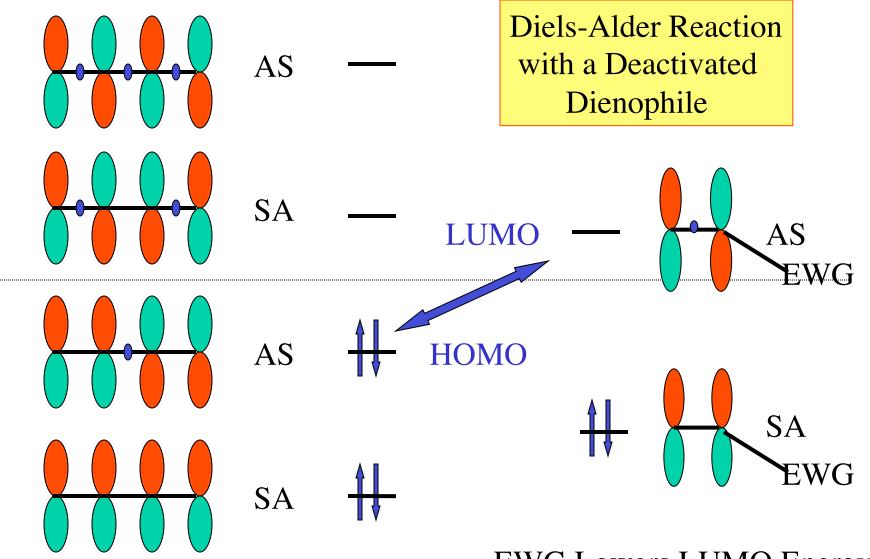
**O**rbital

bonding M.O. -  $\pi_1$ 



SA

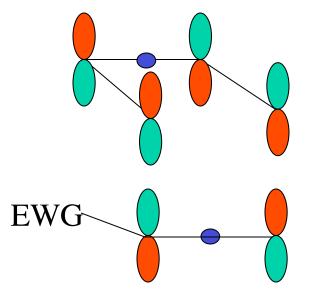




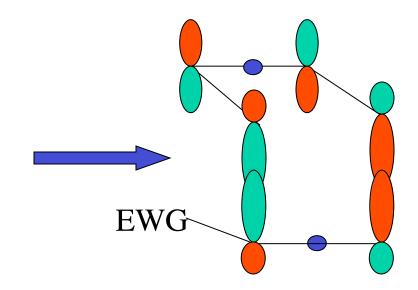
EWG Lowers LUMO Energy

# The Diels-Alder Reaction Transition State

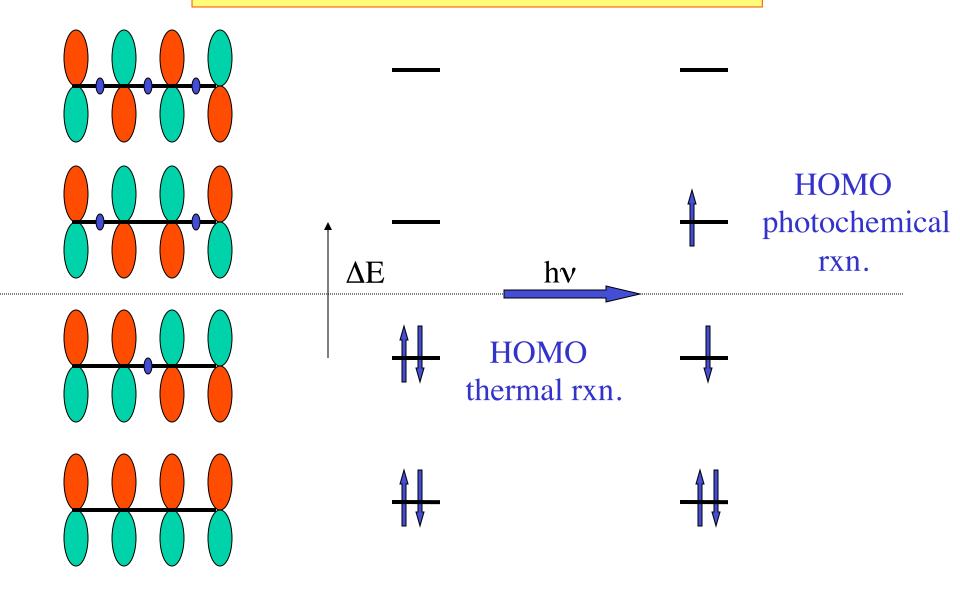
Diene HOMO  $(\pi_2)$ 



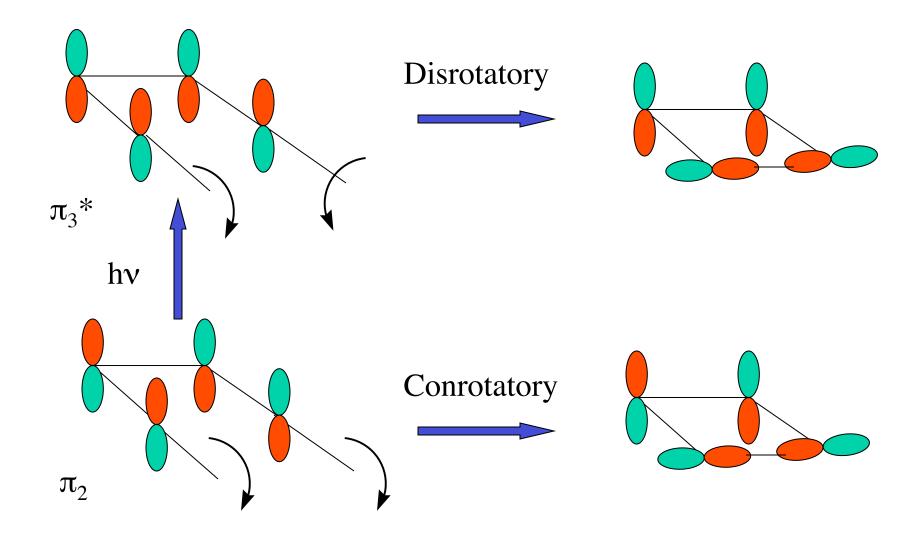




### Photochemical Excitation in 1,3-Butadiene



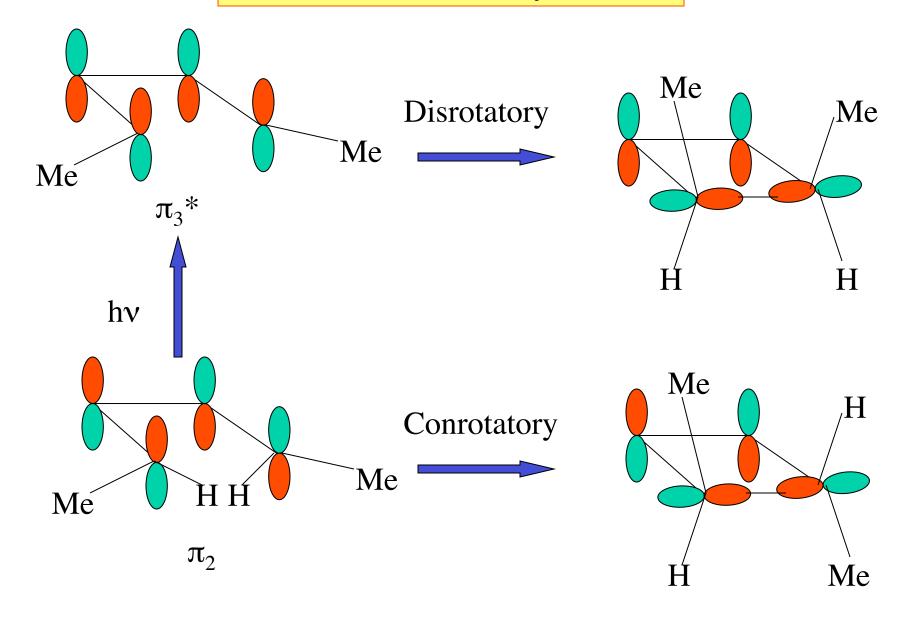
## 1,3-Butadiene — Cyclobutene



but one cyclobutene is just like another cyclobutene

enter stereochemistry

## 1,3-Butadiene — Cyclobutene



### Stereocontrol in the 1,3-Diene Cyclization

