- 1. Arrange the compounds of each set in order of reactivity toward (i) $S_N 2$ displacement and (ii) E2 elimination by KOH in ethanol.
 - (a) 2-bromo-2-methylbutane, 1-bromopentane, 2-bromopentane
 - (b) 1-bromo-3-methylbutane, 2-bromo-2-methylbutane, 2-bromo-3-methylbutane
 - (c) 1-bromobutane, 1-bromo-2,2-dimethylpropane, 1-bromo-2-methylbutane
- 2. Predict the organic products of the following reactions. Label as major and minor, if multiple products are likely.
 - (a) Isopentyl bromide + NaOEt in EtOH
 - (b) Isobutyl bromide + NaOEt in EtOH
 - (c) Methyl bromide + KO-t-Bu in t-BuOH
 - (d) <u>t</u>-butyl bromide + KOCH₃ in CH₃OH
 - (e) 2-bromo-3-methylbutane + NaOEt in EtOH
 - (f) 2-bromo-3-methylbutane + KO-t-Bu in t-BuOH
- 3. Provide reasonable mechanisms to account for the following observations.

- 4. Starting with isopentane, devise efficient syntheses of (a) 2-methyl-2-butene and (b) 2-methyl-1-butene. Any common reagents may be used.
- 5. Explain the following observations.