

**EXAM 3**  
**CHEMISTRY 220a**  
Friday, November 5, 2004

NAME (print): \_\_\_\_\_

TA:\_\_\_\_\_ Day:\_\_\_\_\_ Time:\_\_\_\_\_

Take a few moments to look over the exam. Answer each question on the exam paper.

Important clues, points, and structures are in **bold**.

Do all **preliminary** drawing or computations on the work sheets at the end of the exam. The work sheets will not be graded

The exam is 55 minutes.

**STOP** writing and hand in your exam when you are asked to do so.

**REMEMBER:** Neatness is to your advantage.

1. Structure (25 pts) \_\_\_\_\_

2. Synthesis (25 pts) \_\_\_\_\_

3. Reactions (25 pts) \_\_\_\_\_

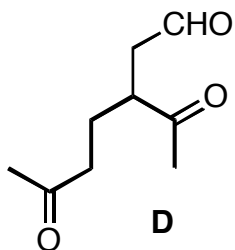
4. Potpourri (25 pts) \_\_\_\_\_

\_\_\_\_\_

Total (100 pts)

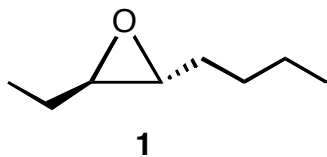


1. **Structure:** (25 pts.) Compound (*S*)-**A** ( $C_{10}H_{16}$ ) reacts with  $H_2$  in the presence of Pd to form achiral **B** ( $C_{10}H_{20}$ ) and achiral **C** ( $C_{10}H_{20}$ ). Ozonolysis and dimethyl sulfide reduction of **A** affords the single enantiomer **D** and formaldehyde. What are the structures **A-D**? Explain and illustrate. Be certain to illustrate the absolute stereochemistry for **A** and **D**.



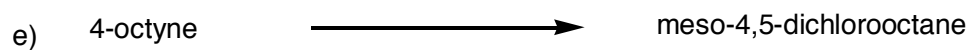
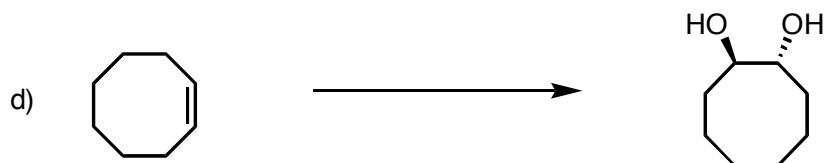
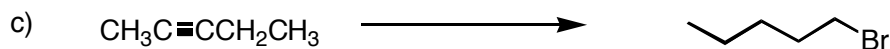
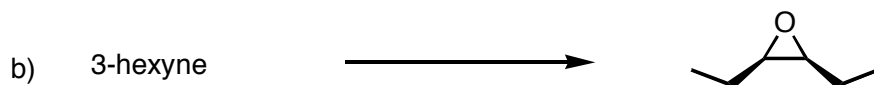
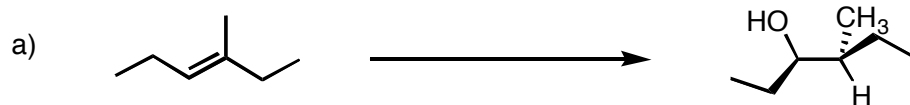


2. **Synthesis:** (25 pts.) A chemist requires a sample of ( $\pm$ )-epoxide **1**. She designs and executes a synthesis of the epoxide using 1-butyne as her sole source of carbon. All other reagents are available to her. Show how she may have accomplished her goal.





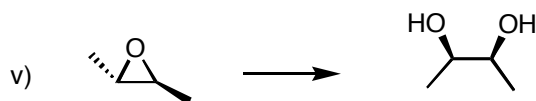
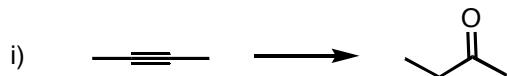
3. **Reactions:** (25 pts.) Provide the reaction conditions in **4 of 5** of the following chemical transformations. Several steps may be required. If you do all five problems, **cross out** the one you do not want graded. **Pay attention to stereochemistry.**





4. **Potpourri:** (25 pts.) Complete **4 of 5** of the following. If you do all five problems, **cross out** the one you do not want graded.

a) Provide (over the arrow) the **catalytic agent** in each of the following single-step reactions.



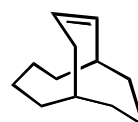
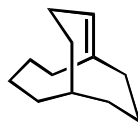
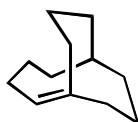
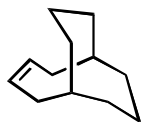
b) **Circle** the reagents that are expected to give meso compounds with (Z)-5-decene .

$\text{Br}_2$      $\text{Br}_2/\text{H}_2\text{O}$     peracid     $\text{OsO}_4/\text{H}_2\text{O}_2$      $\text{CH}_2\text{I}_2/\text{Zn}(\text{Cu})$

c) The heat of formation of 2-hexyne is +25.7 kcal/mol. **Circle** the most likely heat of formation of 1-hexyne.

+29.2    +25.7    +25.2    +20.2    -29.2

d) **Circle** the compound(s) in violation of Bredt's Rule.



none  
of  
these

e) (**Circle** all that apply.) Hydroboration and oxidation of 2-methyl-2-butene affords an alcohol that is:

optical-active    racemic    secondary    tertiary    d,l



Name: \_\_\_\_\_

## Work Sheets



Name: \_\_\_\_\_

7

## Work Sheets



Name: \_\_\_\_\_

## Work Sheets